INDIAN SCHOOL SOHAR FORMATIVE ASSESSMENT - 2

	SUBJECT: SCIENCE
CL	ASS: VI
DA	TE: 05-06-2014
Ge	neral instructions
>	Question nos.1-5 are MCQs carrying 1 mark each.
	<i>Question nos.6-9 are very short answer type carrying 1 mark each.</i> <i>Question nos.10-12 are short answer type carrying 2 marks each.</i>
	Question no.13 is long answer type carrying 5 mark.
1.	Which one is not the example of mixture?
	(a) air (b)salad (c)soil
2.	This liquid is also known as miracle liquid
	(a) Soft drink (b) water (c) oil
3.	Separation of two miscible liquids by distillation i
	difference in their
	(a)melting point (b)boiling point (c)freezing
4.	Solubility of solids with increase in
	(a) increases (b) decreases (c)sometimes
	decreases
5.	Fizzy drinks are example of
	(a) liquid-liquid mixture (b) liquid
	(c) liquid -gas mixture (d) gas -g
6.	Name any two methods that can be used to se
	solids?
7.	Explain the terms filtrate & residue?
8.	Why is loading used to separate suspended particles?
9.	What is sublimation? Give 2 examples of sublimate s
10	Differentiate between
	(a) Solute and Solvent (b) Sedimentation
11	How would you select a method of separation for a

salt.

MAX. MARKS: 20 TIME: 40 Mins.

>	Question nos.10-12 are short answer type carrying 2 marks each.	
	Question no.13 is long answer type carrying 5 mark.Which one is not the example of mixture?	
1.	(a) air (b)salad (c)soil (d)water	
2	This liquid is also known as miracle liquid	
	(a) Soft drink (b) water (c) oil (d) sugar solution	
3.	Separation of two miscible liquids by distillation is done based on the	
	difference in their	
	(a)melting point (b)boiling point (c)freezing point (d)both a and b.	
4.	Solubility of solids with increase in temperature	
	(a) increases (b) decreases (c)sometimes increases (d)sometimes	
	decreases	
5.	Fizzy drinks are example of	
	(a) liquid-liquid mixture (b) liquid -solid mixture	
	(c) liquid -gas mixture (d) gas -gas mixture	
6.	Name any two methods that can be used to separate solids from the	
7	solids?	
7.	Explain the terms filtrate & residue?	
8.	Why is loading used to separate suspended particles?(write two points)	
9.	What is sublimation? Give 2 examples of sublimate substances.	
10	.Differentiate between	
	(a) Solute and Solvent (b) Sedimentation and decantation	
11	.How would you select a method of separation for any mixture? (Write three	
	points)	
12.What do you mean by saturated solution of sugar? What will happen when		
12		
	this solution is heated?	
13	.Describe an experiment to show how to separate a mixture of saw dust &	